

Lean blowoff of kerosene swirl spray flames using LES-CMC

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UKCTRF Annual Meeting

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Current Research Focus

- Current emphasis on fuel-lean-combustion designs to reduce emissions
 - Lean blowoff (LBO) is a critical factor
 - Lean designs increase instabilities
- Alternative jet fuels are being developed with dramatically different and unique compositions
- Fuel-chemistry observed to have largest effect at lean conditions^[1]

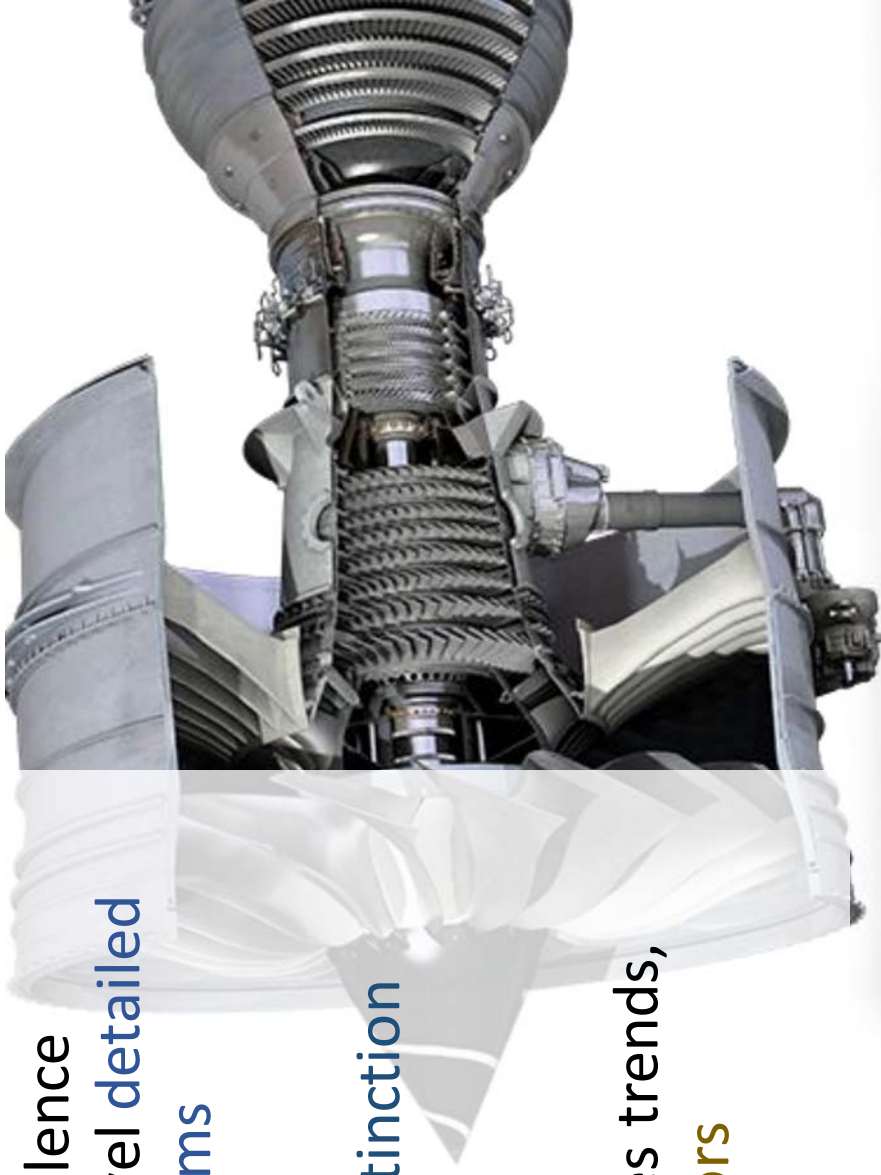


SR71 – afterburner blowout



Research Objectives

- Evaluate the **LES-CMC** turbulence combustion model using novel detailed **kerosene chemical mechanisms**
- Model **lean blow-off** and **extinction** phenomena
- Analyze intermediate species trends, particularly for **soot precursors**



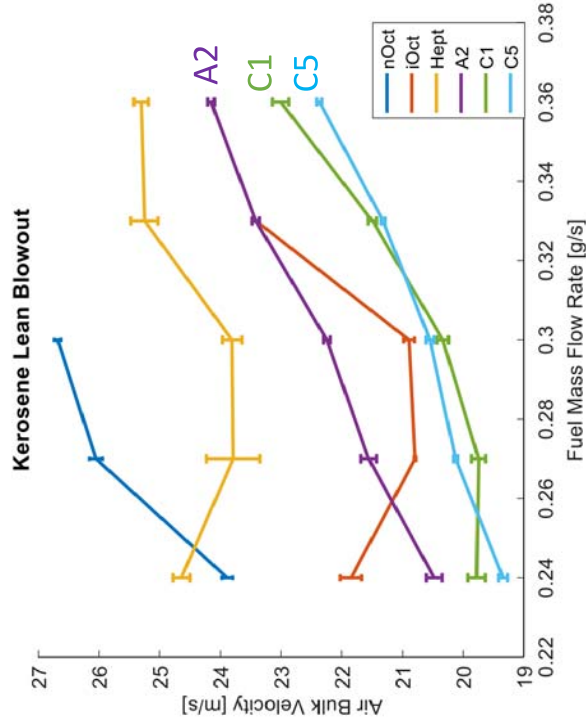
Outline

1. Context and Objectives
- 2. Setup and Methods**
3. LES-CMC Results
 - Comparison with Experiments
 - Global blow-off
 - Local extinctions
4. Conclusions
5. Current/Future Work

Setup – Relevant Literature

Experiments:

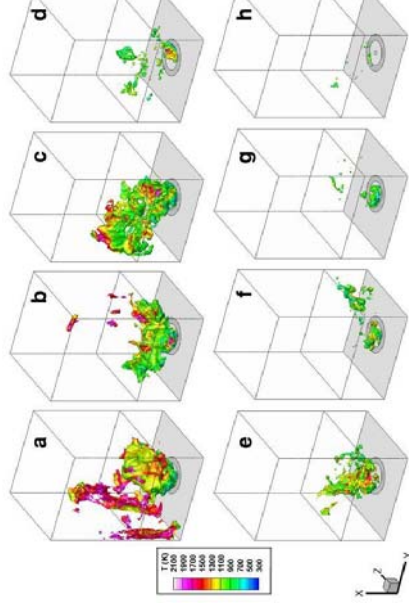
- Sidey et al. 55th AIAA Meeting. (2017)
- Allison et al. AIAA Proceedings. (2018)



Allison et al. AIAA Proceedings. 2018.

Simulations involving blow-off:

- Tyliczszak, Cavaliere & Mastorakos. Flow Turb. Comb. (2014)
 - Blow-off of n-heptane swirl flame using LES-CMC
- Zhang & Mastorakos. Flow Turb. Comb. (2016)
 - Blow-off of methane swirl flame using LES-CMC
- Giusti & Mastorakos. PROCI. (2017)
 - Blow-off of ethanol swirl spray flame using LES-CMC



Zhang & Mastorakos. FTC. 2016.

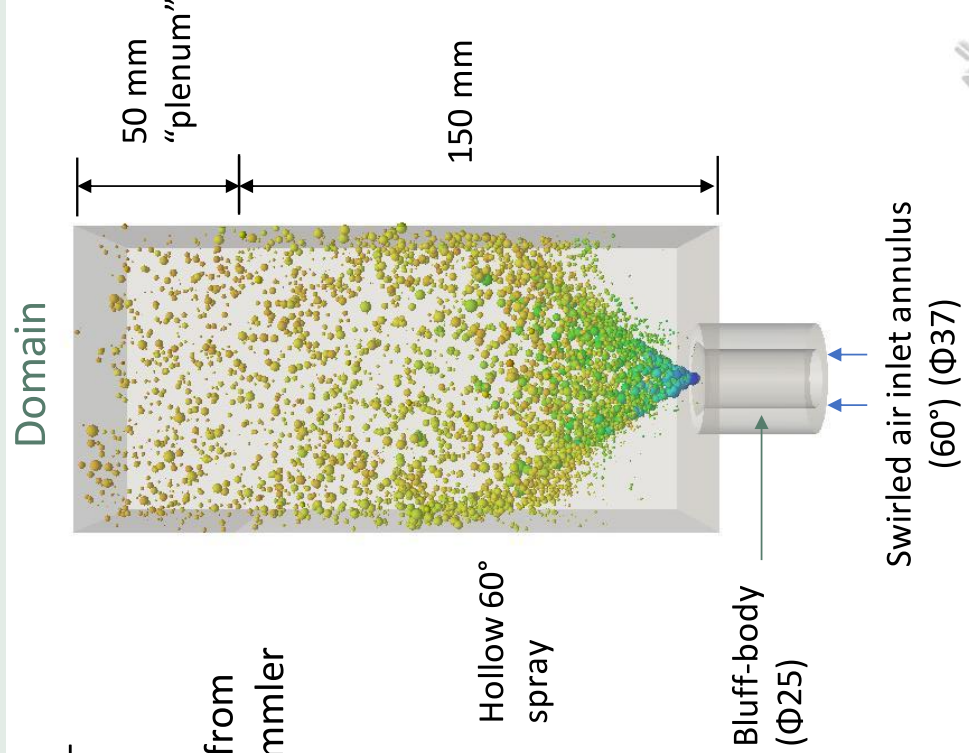
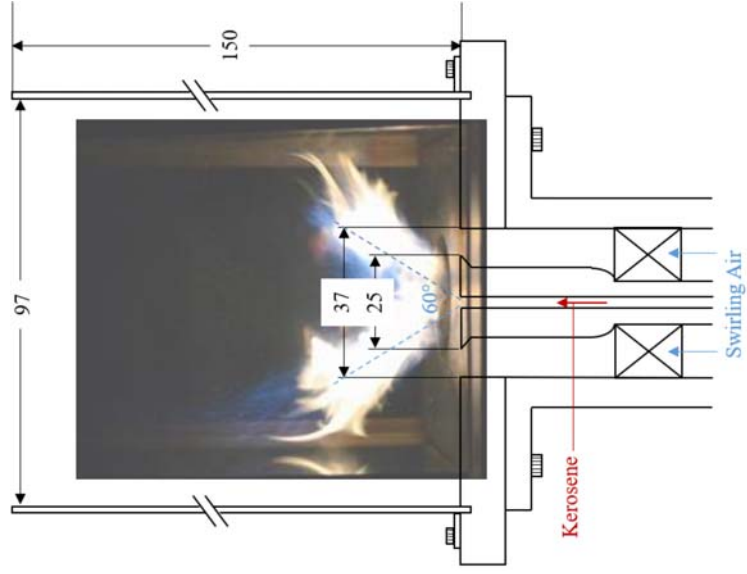
Burner Setup

- Cambridge bluff body non-premixed swirl-stabilized burner
- Hollow cone kerosene spray was injected from centre of bluff body at 60° using Rosin-Rammler distribution with SMD of 60 μm based on experiments [1-2]

Kerosene Fuel:

- ❖ A2: conventional Jet A, $C_{11}H_{22}$

Hollow 60° spray

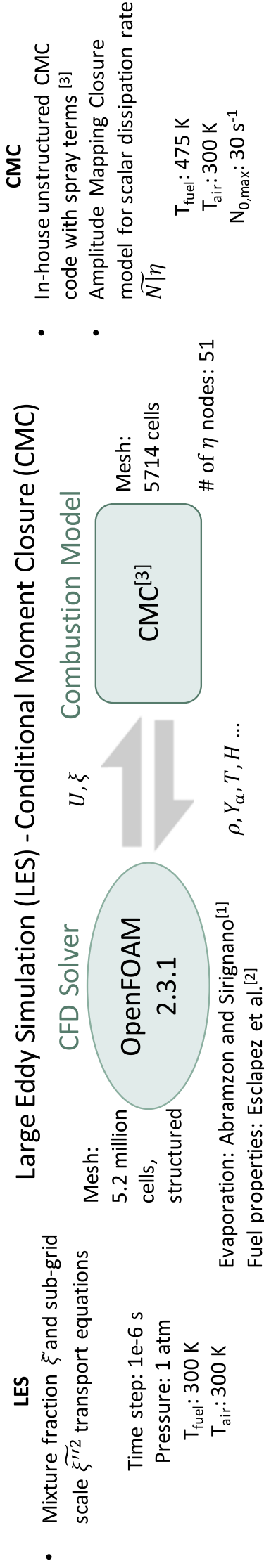


Flow conditions

Φ_{overall}	U_b Air [m/s]	$\dot{m}_{\text{fuel}}$ [g/s]
0.36	15.85	0.27
0.25	22.7	0.27



Numerical Methods and Chemical Mechanism



HyChem “A2” kerosene mechanism^[5]: lumped pyrolysis, single-component fuel, 119 species, 843 reactions

- Main products of pyrolysis are: ethylene, propene, iso-butene, 1-butene, benzene, toluene, hydrogen and the methyl radical
- Oxidized with detailed mechanism USC-Mech II (not pictured)

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POSF10325 => 1.7426762C2H4 + 0.8190578C3H6 + 0.0871338iC4H8 + 0.2614014C4H81 + 0.1700C6H6 + 0.1633333C6H5CH3 + 0.5H + 1.5CH3
POSF10325+H => H2 + 0.45CH4 + 1.5945764C2H4 + 0.7494509C3H6 + 0.0797288iC4H8 + 0.2391865C4H81 + 0.2465C6H6 + 0.2368333C6H5CH3 + 0.3H + 0.7CH3
POSF10325+CH3=> 1.45CH4 + 1.5945764C2H4 + 0.7494509C3H6 + 0.0797288iC4H8 + 0.2391865C4H81 + 0.2465C6H6 + 0.2368333C6H5CH3 + 0.3H + 0.7CH3
POSF10325+OH => H2O + 0.45CH4 + 1.5945764C2H4 + 0.7494509C3H6 + 0.0797288iC4H8 + 0.2391865C4H81 + 0.2465C6H6 + 0.2368333C6H5CH3 + 0.3H + 0.7CH3
POSF10325+O2 => H2O2 + 0.45CH4 + 1.5945764C2H4 + 0.7494509C3H6 + 0.0797288iC4H8 + 0.2391865C4H81 + 0.2465C6H6 + 0.2368333C6H5CH3 + 0.3H + 0.7CH3
POSF10325+H02=> H2O2 + 0.45CH4 + 1.5945764C2H4 + 0.7494509C3H6 + 0.0797288iC4H8 + 0.2391865C4H81 + 0.2465C6H6 + 0.2368333C6H5CH3 + 0.3H + 0.7CH3
POSF10325+O => OH + 0.45CH4 + 1.5945764C2H4 + 0.7494509C3H6 + 0.0797288iC4H8 + 0.2391865C4H81 + 0.2465C6H6 + 0.2368333C6H5CH3 + 0.3H + 0.7CH3

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[1] Abramzon, B., & Sirignano, W. Int. J. Heat Mass Transfer, 32, (1989).

[2] Esclapez, L. et al., Combust. Flame, 181, (2017).

[3] Giusti, A. and Mastorakos, E., Proc. Combust. Inst., 36, (2017).

[4] Perini, F. et al., Energy and Fuels, 26, (2012)

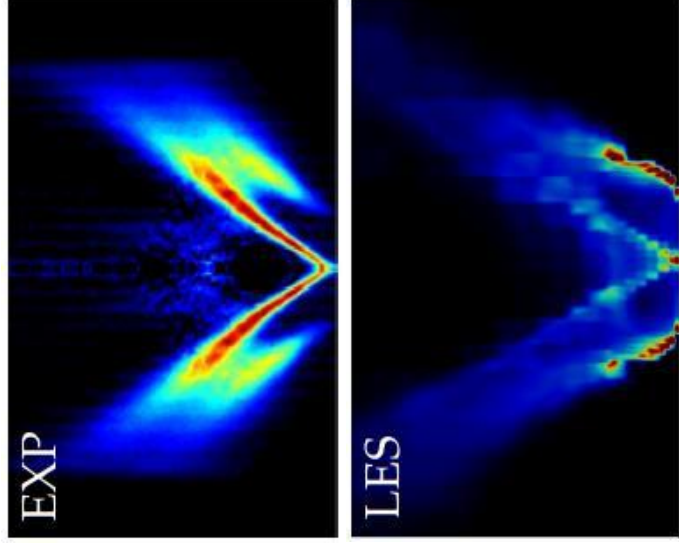
[5] Wang, H. et al. Combust. and Flame, 193 (2018).

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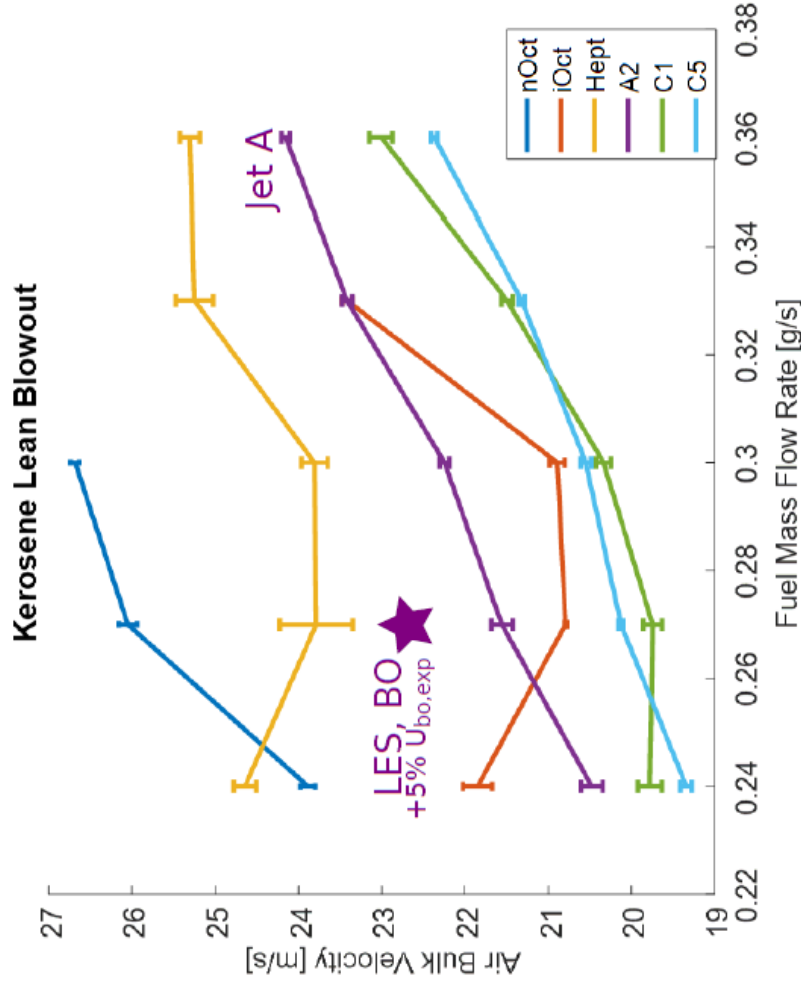


Results – Comparison with experiment



Inverse Abel-transformed time-averaged OH*

Time-averaged 2D cut-plane of OH*

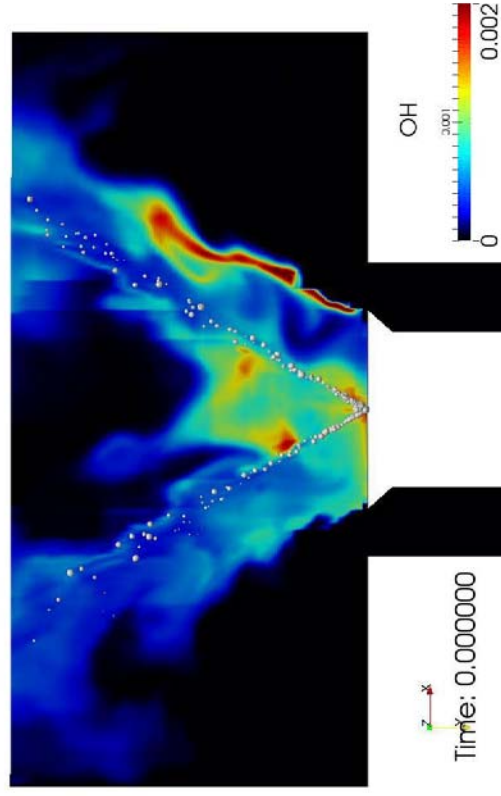


- [1] Sidey, J.A.M. et al., 2017, 55th AIAA Aerospace Sciences Meeting.
 [2] Allison, P. M et al., *AIAA Sci-Tech 2018*, , No. AIAA 2018-1421, 2018.



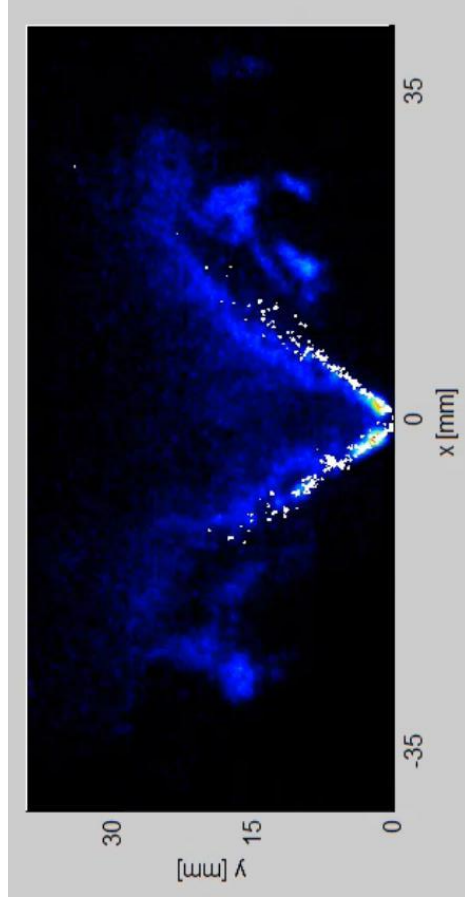
Results – Comparing with experiment

Simulation



LES-CMC of Lagrangian
spray and gaseous OH

Experiment



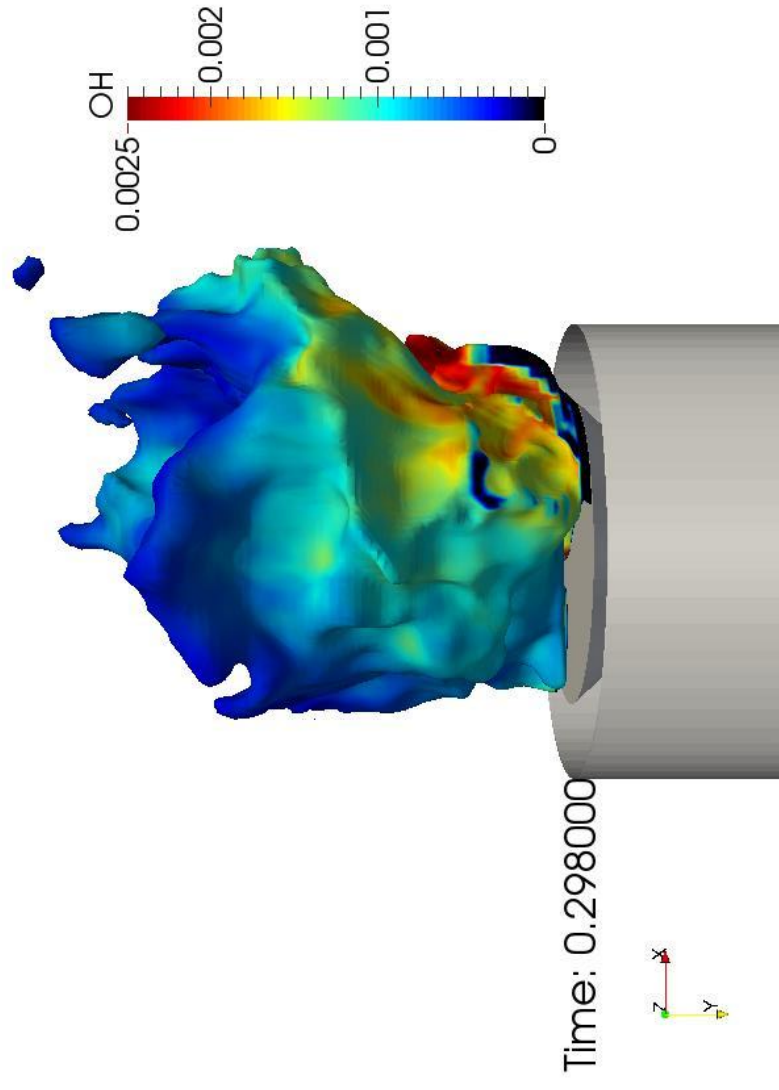
OH-PLIF Mie of Jet-A [1]



[1] Sidey, J.A.M. et al., 2017, 55th AIAA Aerospace Sciences Meeting.



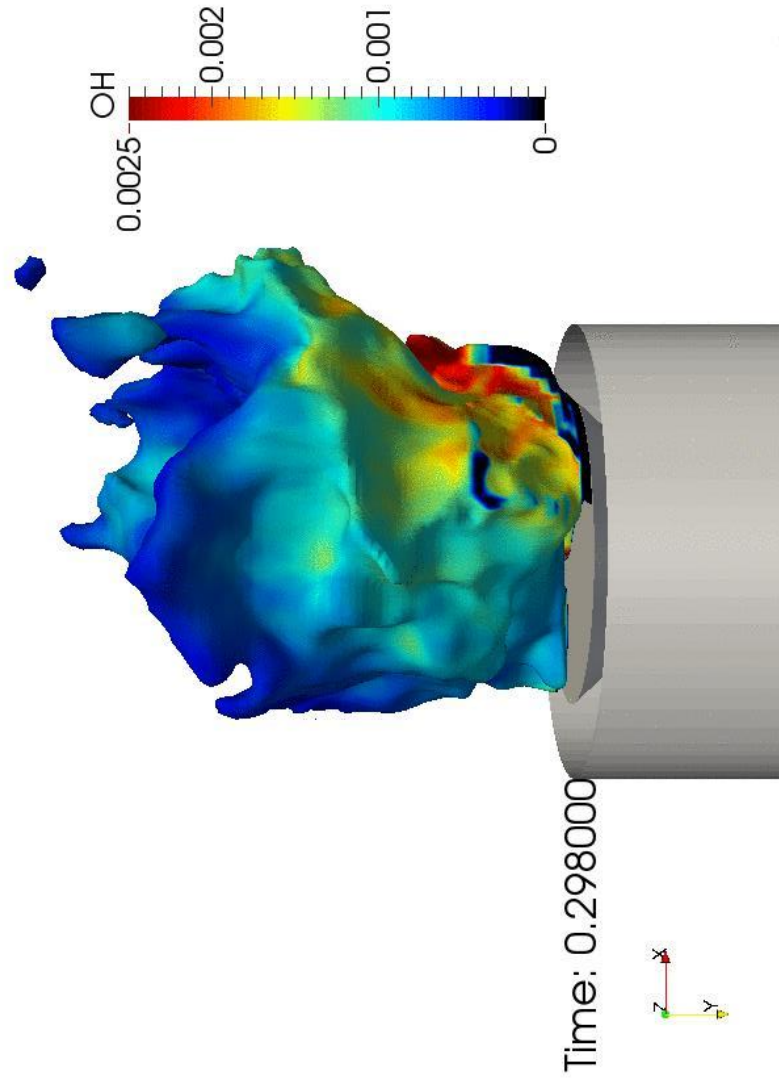
Results – Global blowoff



ξ_{st} isosurface colored
with OH mass fraction



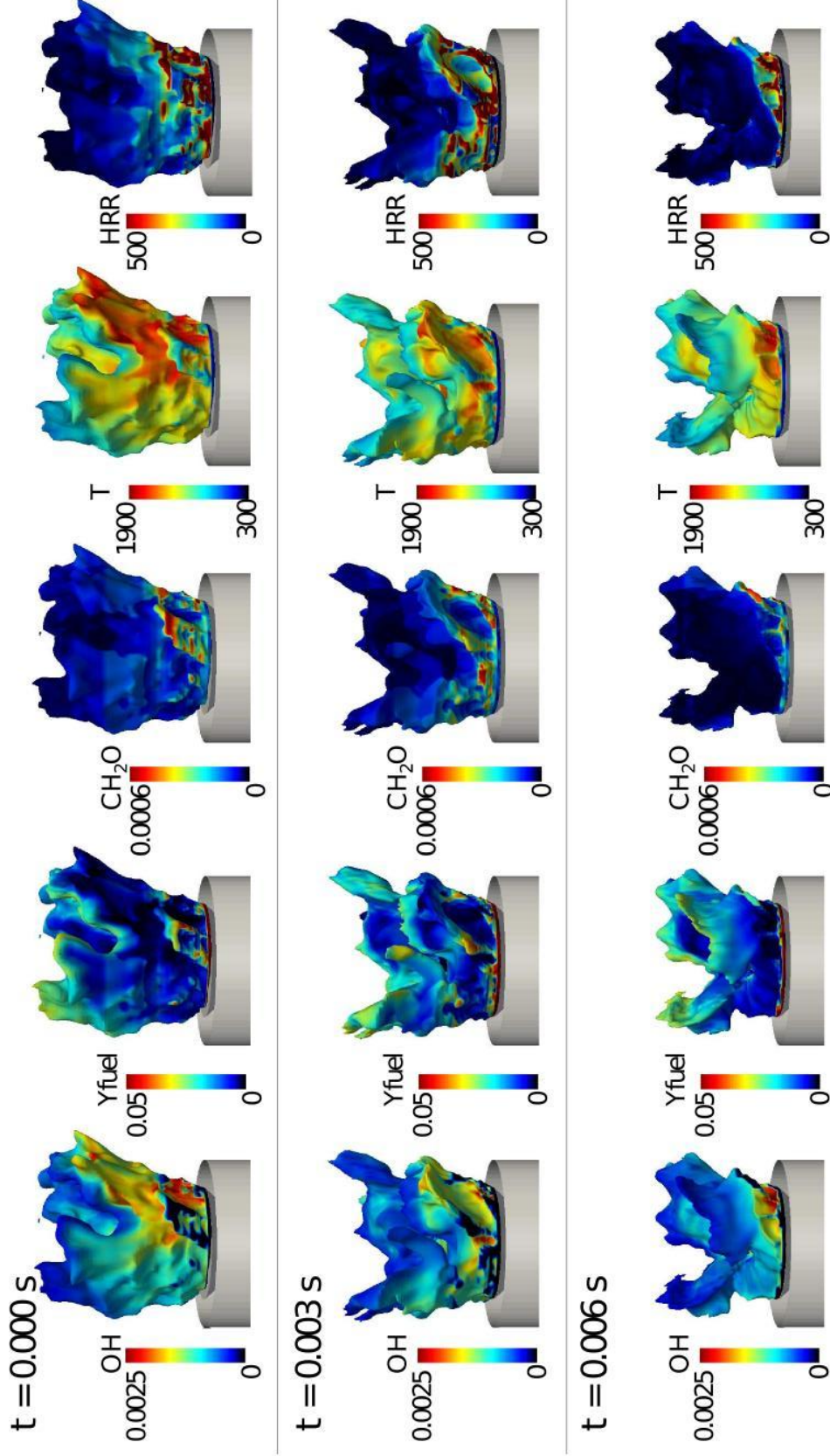
Results – Global blowoff



ξ_{st} isosurface colored
with OH mass fraction

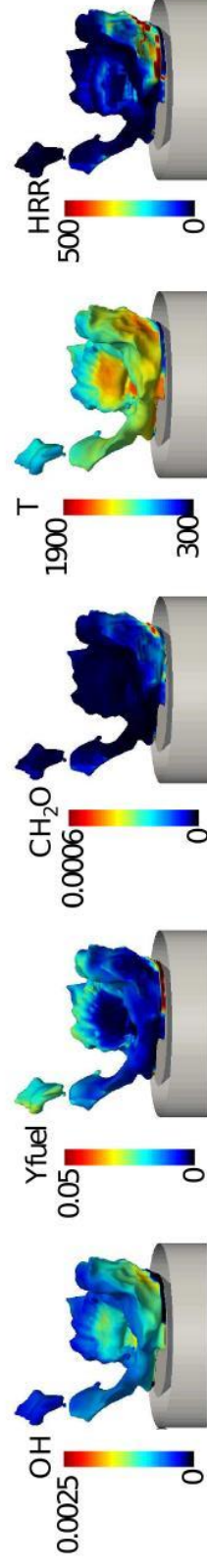


Results – Global blowoff

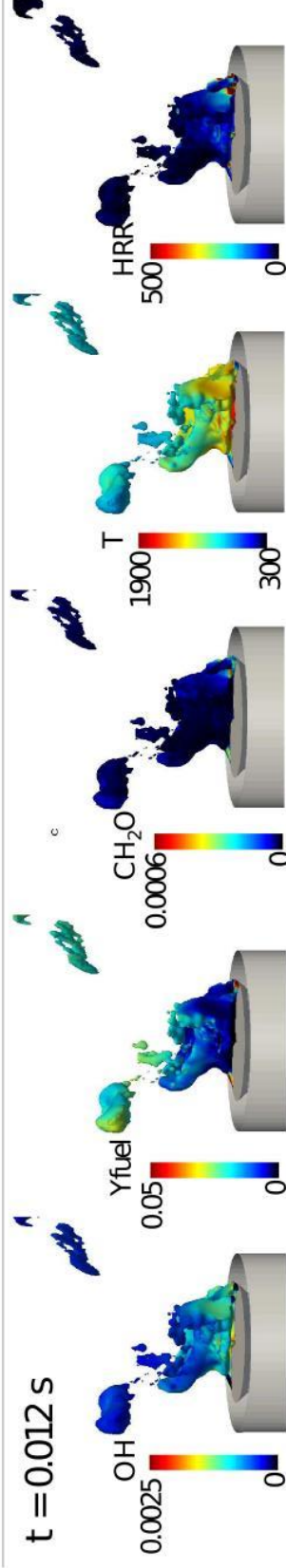


Results – Global blowoff

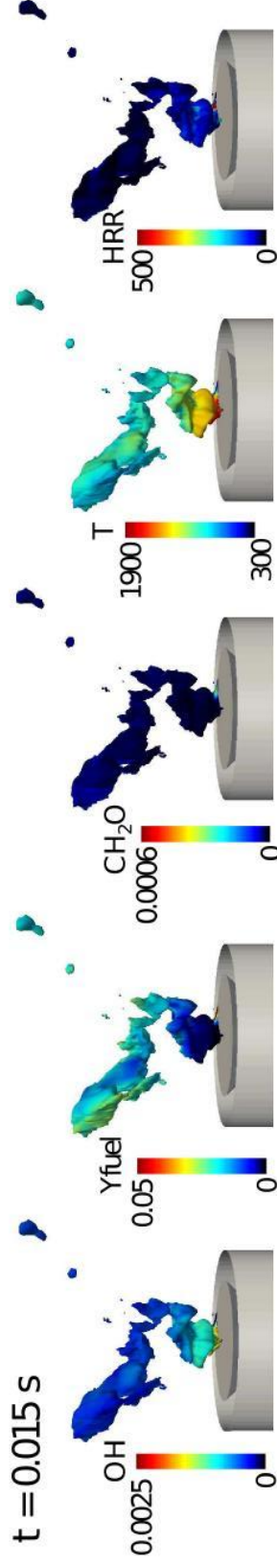
$t = 0.009$ s



$t = 0.012$ s



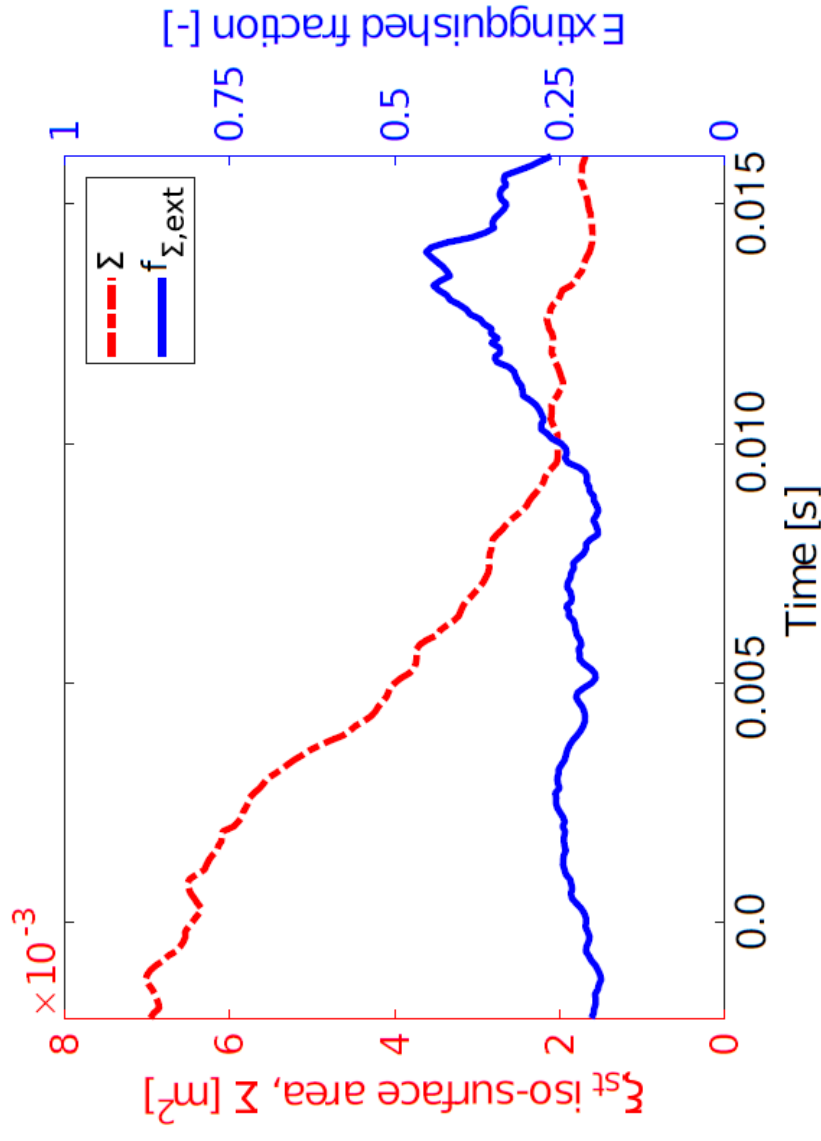
$t = 0.015$ s



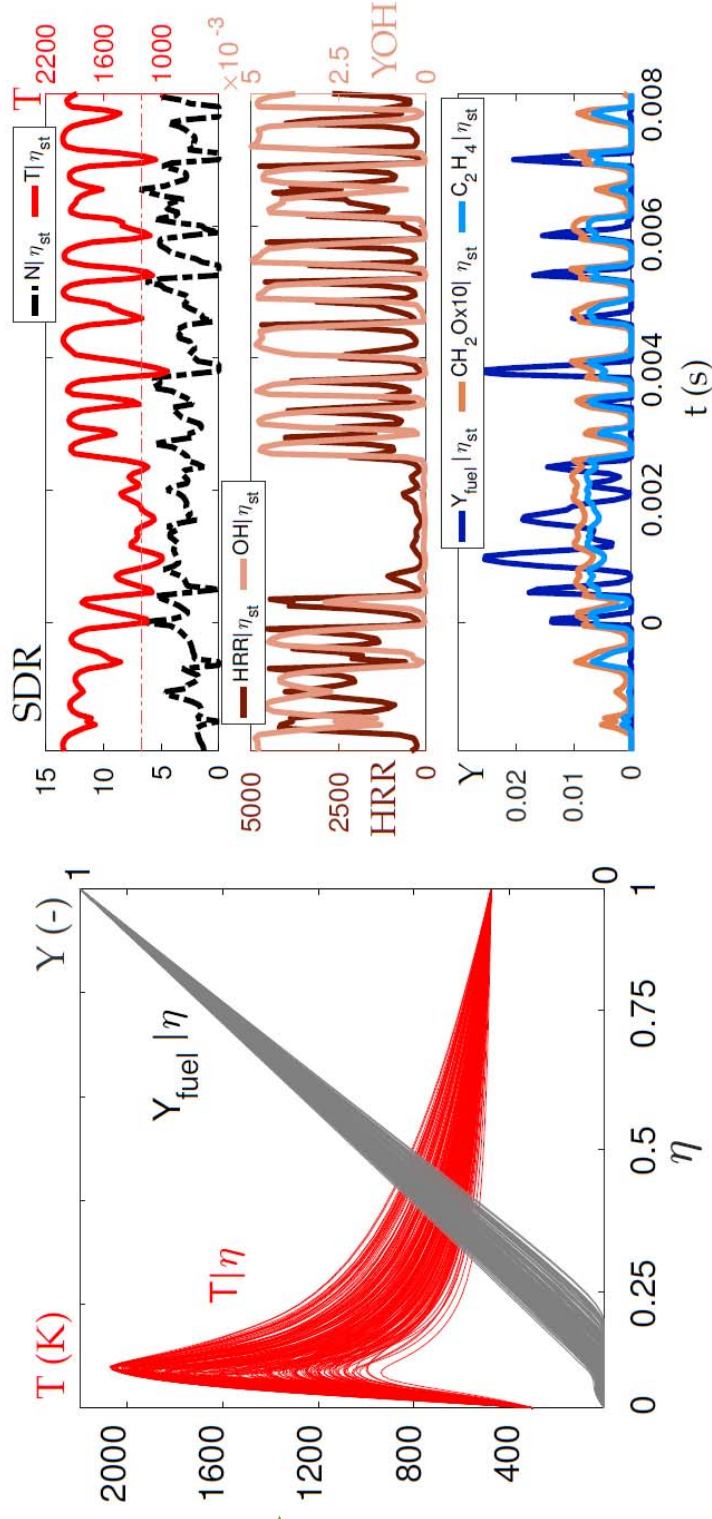
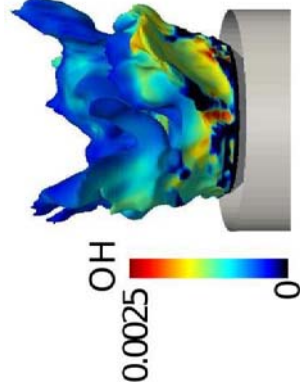
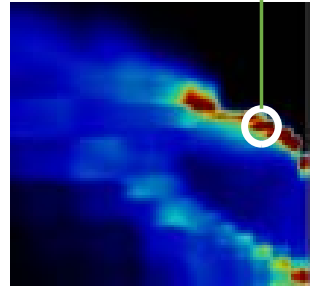
Results – Extinguished fraction

$$f_{ext} = \frac{\Sigma |(\widehat{Y_{OH}})_{\xi_{st}} < 0.00045|}{\Sigma} \quad [1]$$

- The extinguished fraction, f_{ext} :
- $f_{ext} = 1$ is complete extinction
- $f_{ext} = 0$ is fully burning

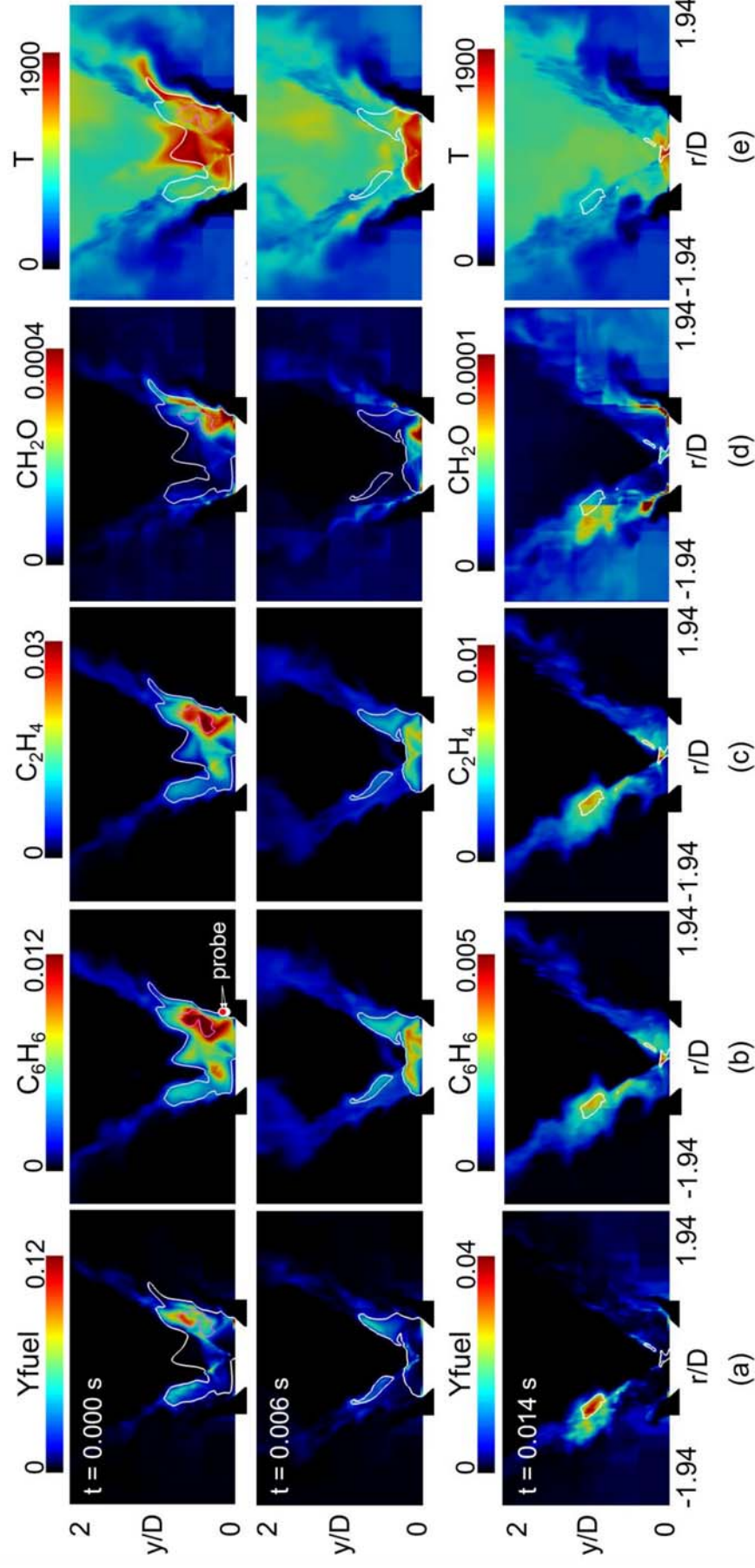


Results – Local extinction

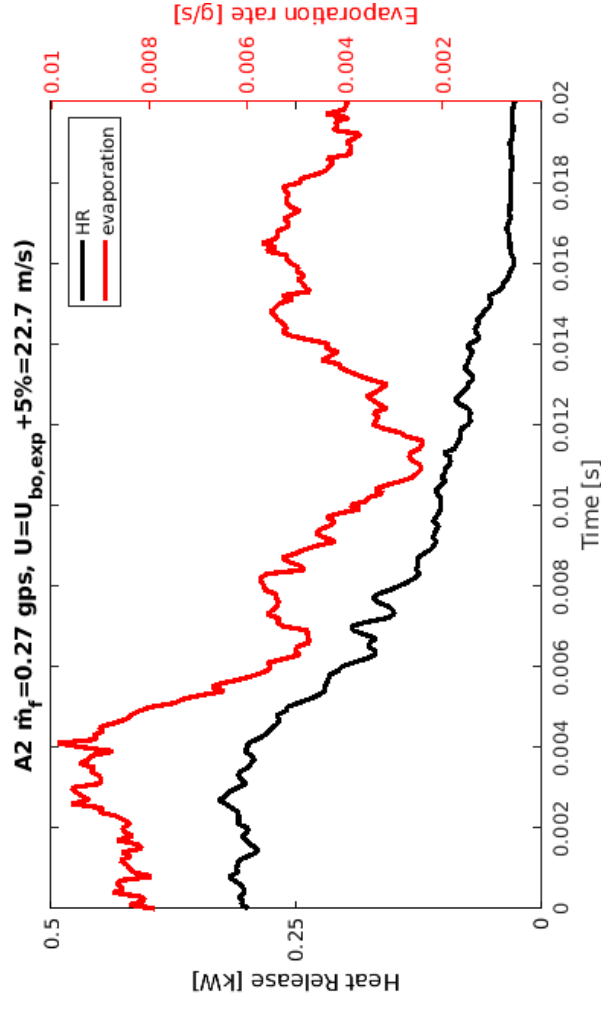
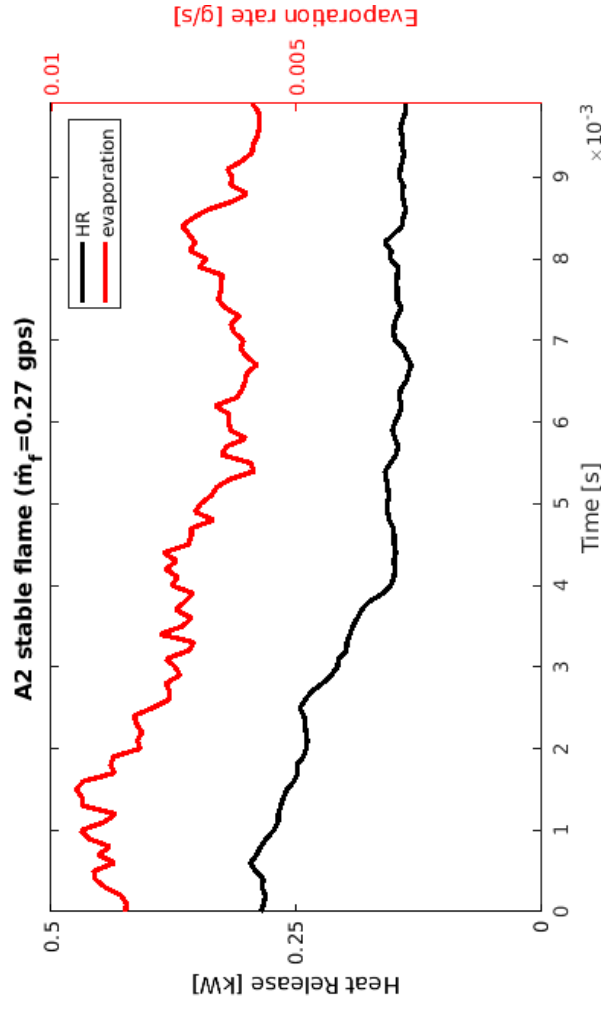


- Probe of CMC cell along the shear layer showing quantities in mixture fraction space

Results – Intermediate species



Results – Heat release and evaporation rate



Local volume integrated heat release rate and evaporation using volume of cells around the stoichiometric mixture fraction isosurface

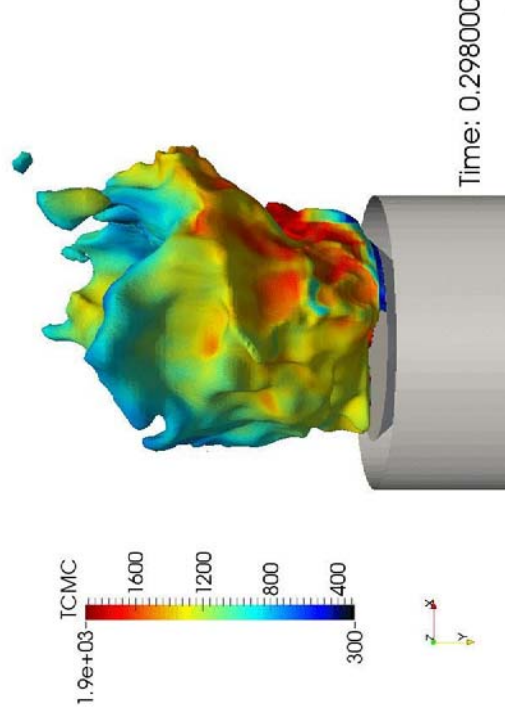
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Conclusions

- Simulated the blowoff transient of a Jet-A swirl spray flame using detailed chemistry
- Captured local extinctions visually and quantitatively
- Flame appeared to blowoff because of both fuel starvation and local extinctions
- Observed blowoff of the flame within +5% of experimental air bulk velocity
- Affirmed capability of LES-CMC to model blowoff of real fuels

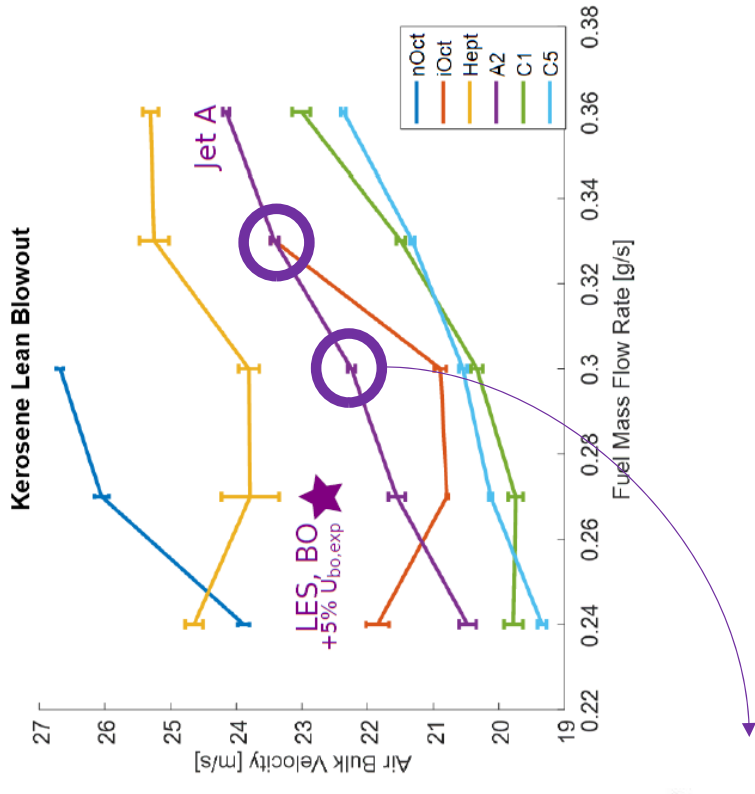
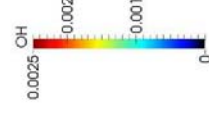
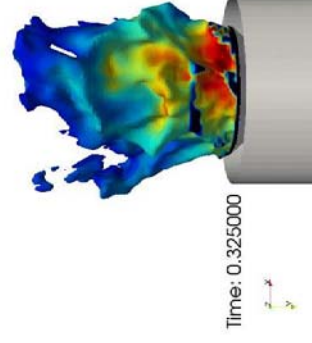


Current & Future Work

- Two more mass flow rate conditions from Jet-A blowoff curve being simulated

➤ Want full simulated blowoff curve for kerosene

- Simulation of alternative jet fuel C1 from National Jet Fuels Combustion Program, comparison with Jet-A and experiments



+15% of experimental blowoff velocity



Acknowledgments

We would like to thank the following for their support with this work:



Rolls-Royce



UK CONSORTIUM
ON TURBULENT
REACTING FLOWS



Dr M. Philip Sitte (Siemens)

Prof Hai Wang (Stanford) & Prof Fokion Egolfopoulos (USC)

Thank you for listening!

